

A FACILE SYNTHESIS OF α,β -UNSATURATED β -BROMO-N-ACYL- α -AMINO ACIDS

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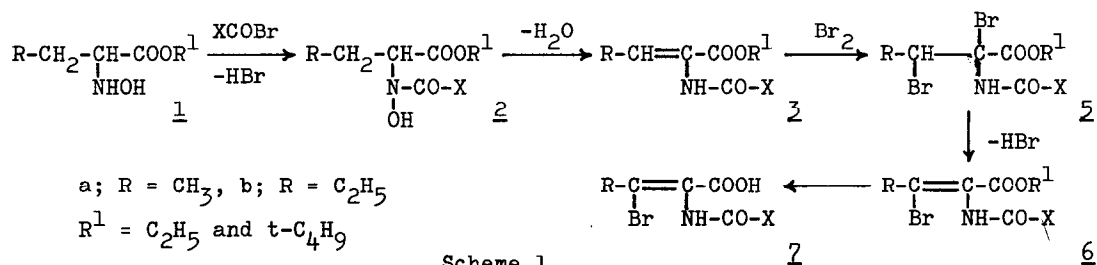
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(Received in Japan 12 December 1973; received in UK for publication 2 January 1974)

In previous papers,¹⁻³ we reported the synthesis of α,β -unsaturated N-acyl- α -amino acid ester (3) by the condensation of ethyl α -oxocarboxylate with amide or by the treatment of N-acetoxy-N-acyl- α -amino acid ester with Et₃N. Up to date, however, the available synthetic method of the title compounds 6 and 7 has never been reported except the preparation of β -halo-sec- and tert-enamines.⁴⁻⁶ Owing to the synthetic utility of 6 and 7, a facile synthetic method was pursued here.

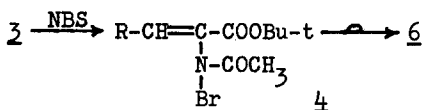
Attempts on addition of Br₂ to 3 (R¹=Et) in CHCl₃ and subsequent elimination of HBr was unsuccessful except the direct conversion of 3a (R¹=t-Bu) into 7a (X=CH₃; yield 59%. mp 150°C; X=OEt; yield 60%. mp 154-156°C; X=phthaloyl; yield 38%. mp 188°C). Furthermore, it was found that the treatment of ethyl α -hydroxyamino-propanoate (1a) with bromoacetyl bromide in benzene in the absence of pyridine⁷ gave directly the expected ethyl N-bromoacetyl- α -amino- β -bromo- α -pentenoate (6a; X=CH₂Br, mp 118-119°C), though the yield was 6.7%. It was deduced that the formation of 3 via the corresponding hydroxamic acid ester (2)² was successively followed by addition of Br₂ formed oxidatively from HBr, and by elimination of HBr to



Scheme 1

afford the compound **6**.

On the other hand, the treatment of **3** with N-bromosuccinimide in CCl_4 gave unstable oily product; tert-butyl N-acetyl-N-bromo- α -alkenoate (**4a**; ν_{max} 1740, 1720, 1680 cm^{-1} ; **4b**; ν_{max} 1740, 1725, 1680 cm^{-1}), in quantitative yields, which gradually rearranged at room temperature for two days to give **6** as crystals in good yields. Compound **6** was composed of E- and Z-isomers, which were separated



Scheme 2

chromatography on silica-gel column using benzene-ethyl acetate (5 : 1) as effluent (Table 1). All of the new compounds and their structures have been characterized spectroscopically and gave satisfactory elementary analysis.

Table 1. tert-Butyl N-Acetyl- α -amino- β -bromo- α -alkenoate (**6**)

Compound		Mp $^{\circ}\text{C}^{\text{a}}$	Yield (%)	IR Spectrum, cm^{-1} in KBr	NMR Spectrum, in CDCl_3 , NH
6a	E	138-139	54	3340,1715,1680,1645	7.06
	Z	101.5-102.5	36	3340,1715,1685,1640	7.94
6b	E	84-87	45	3270,1725,1670,1635	7.16
	Z	syrup	18	3270,1730,1680	8.35

a) Recrystallization from dibutyl ether gave colorless needles.

References and Note

- 1) C. Shin, M. Fujii, and J. Yoshimura, *Tetrahedron Lett.*, **1971**, 2499.
C. Shin, K. Sato, A. Ohtsuka, K. Mikami, and J. Yoshimura, *Bull. Chem. Soc. Japan*, in press.
- 2) C. Shin, K. Nanjo, E. Ando, M. Sakurai, and J. Yoshimura, Preprints for the 27th Annual Meeting of the Chemical Society of Japan, Nagoya, October, 1972, Vol. 1, p. 527.
- 3) C. Shin, K. Nanjo, and J. Yoshimura, *Chemistry Lett.*, **1973**, 1039.
- 4) S. J. Huang and M. V. Lessard, *J. Org. Chem.*, **35**, 1204 (1970).
- 5) H. Ahlbrecht and M. T. Reiner, *Tetrahedron Lett.*, **1971**, 4901.
- 6) L. Duhamel, P. Duhamel, and J-M. Poirier, *Tetrahedron Lett.*, **1973**, 4237.
- 7) This reaction performing in the presence of pyridine, the corresponding hydroxamic acid esters were obtained in good yield (reference 2).